

chapter 9 stoichiometry multiple pdf

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In chemistry, the law of multiple proportions is one of the basic laws of stoichiometry used to establish the atomic theory, alongside the law of conservation of mass (matter) and the law of definite proportions. It is sometimes called Dalton's Law after its discoverer, the British chemist John Dalton, who published it in the first part of the first volume of his "New System of Chemical ...

Law of multiple proportions - Wikipedia

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Macromolecule - Wikipedia

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